

Unrecognized conceptual hurdles to understanding evolutionary biology ... and some of their physiochemical roots.

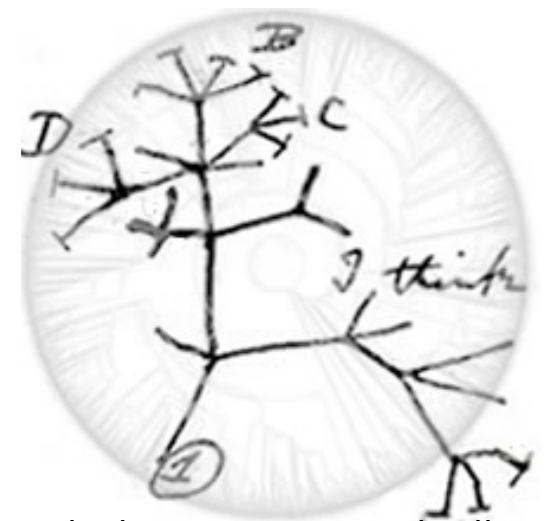
Mike Klymkowsky

Professor, Molecular, Cellular & Developmental Biology

Co-Director, CU 🍏 Teach Science/Math Teacher Recruitment & Certification Program

AAAS Symposium on Discipline Based Science Education

Sunday, 15th February 2009



Charles Darwin & David Hillis

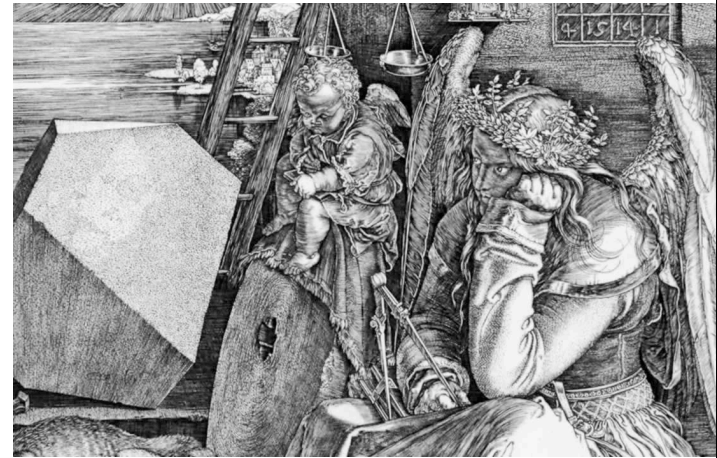
Why are scientific ideas difficult to grasp / apply / accept?



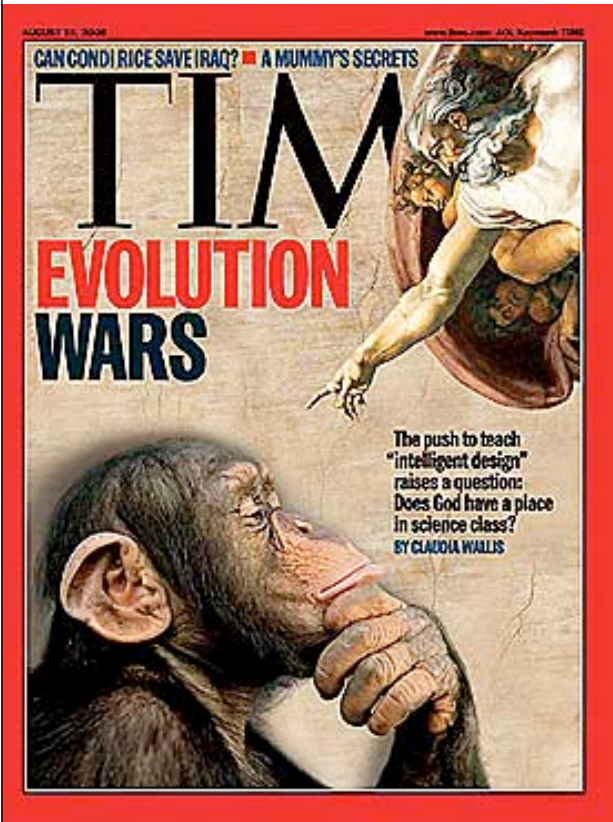
Many are deeply counter-intuitive

These include ...

- The Laws of Thermodynamics
- Atomic theory / quantum mechanics
- Entropy (in general)
- The age (and size) of the universe

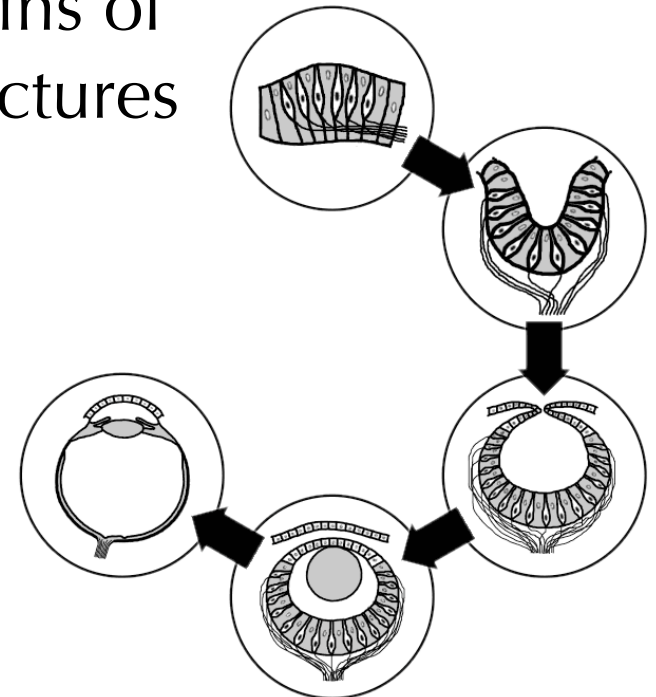


Not to mention
physiochemical basis of life
cell theory of life

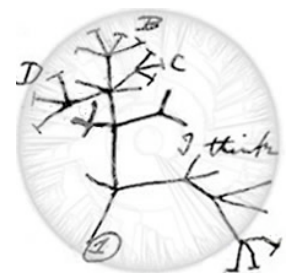
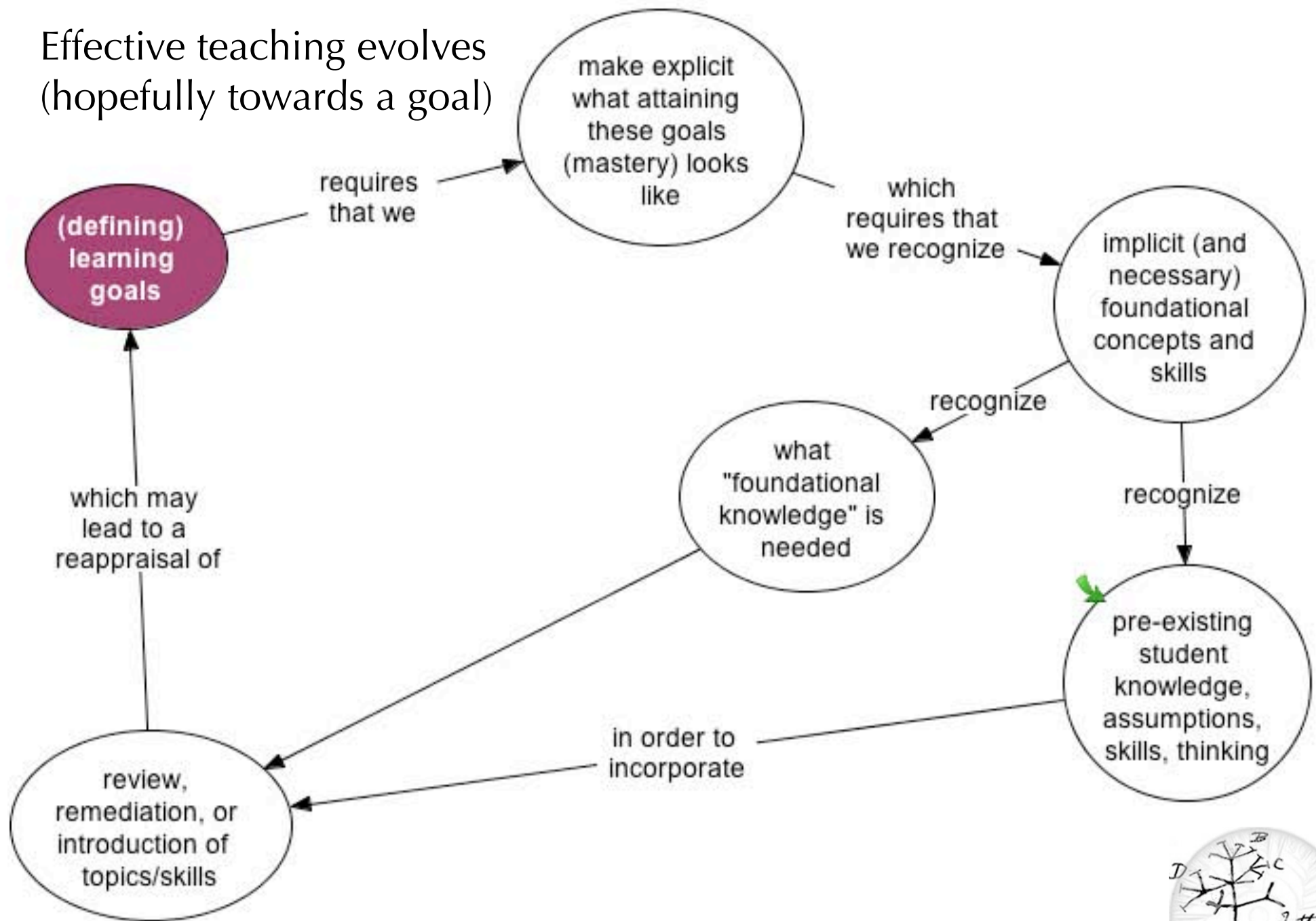


Random origins of complex structures

which can lead to some serious sociocultural ramifications

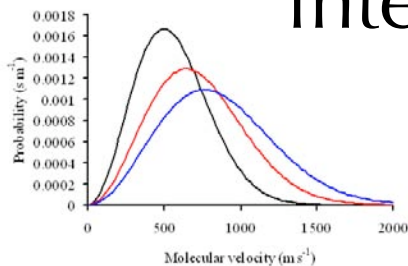


Effective teaching evolves
(hopefully towards a goal)



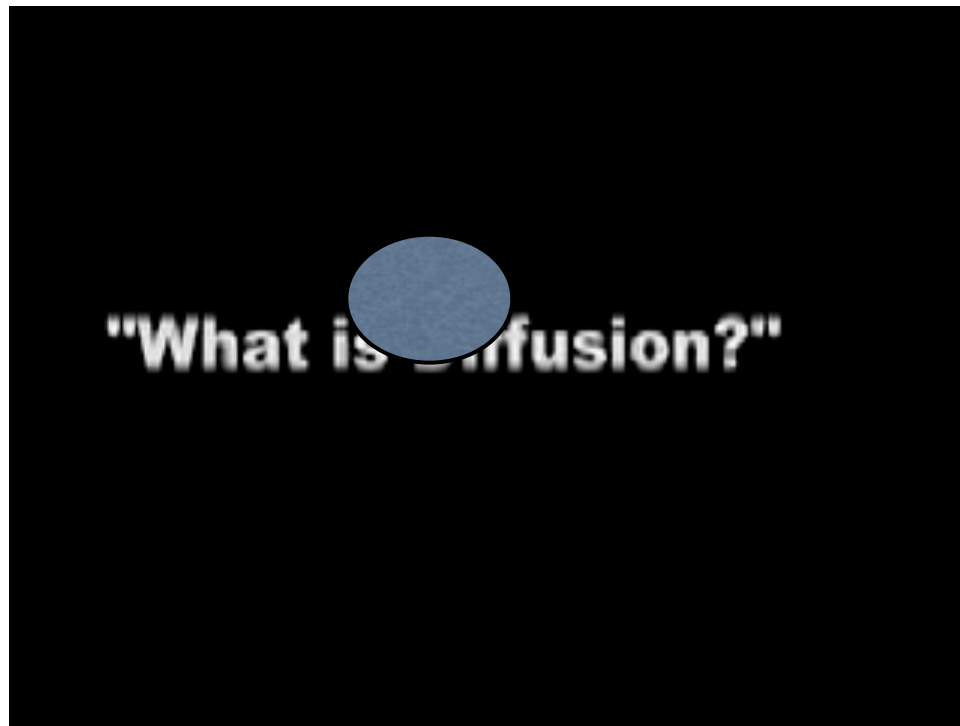
Biology's physiochemical foundations

- Conservation of energy and matter
 - understanding energy and matter
 - when is $e = mc^2$ applicable and when isn't it.
- Laws of Thermodynamics
 - understanding work and entropy and the assembly of complex structures
 - kinetic vs potential energy
 - molecular movement (diffusion)
 - interaction stability, specificity & dynamics



Biology's physiochemical foundations

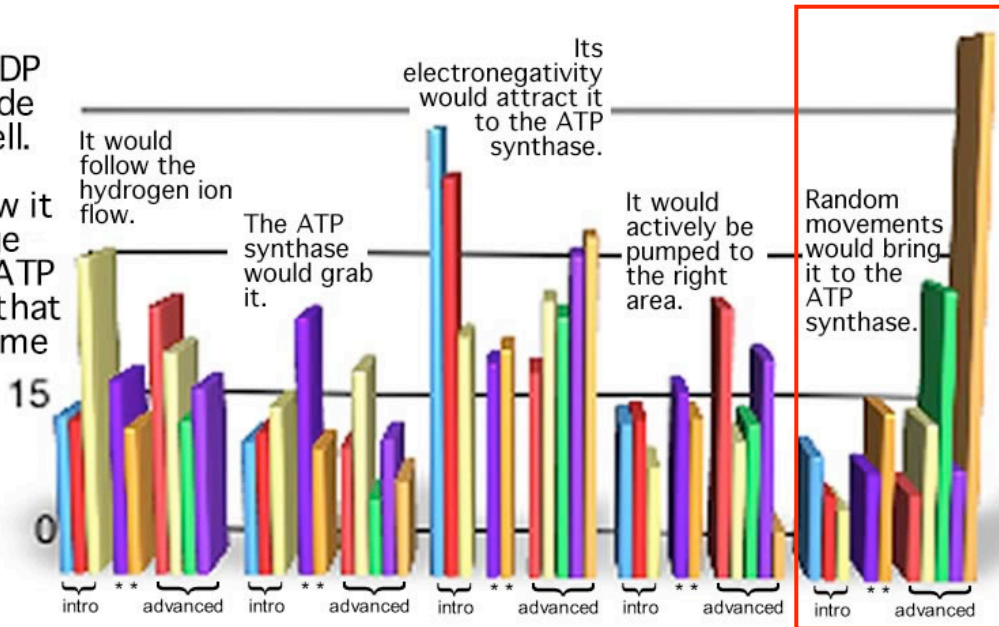
Is there evidence that students have trouble with physiochemical ideas?



yes!



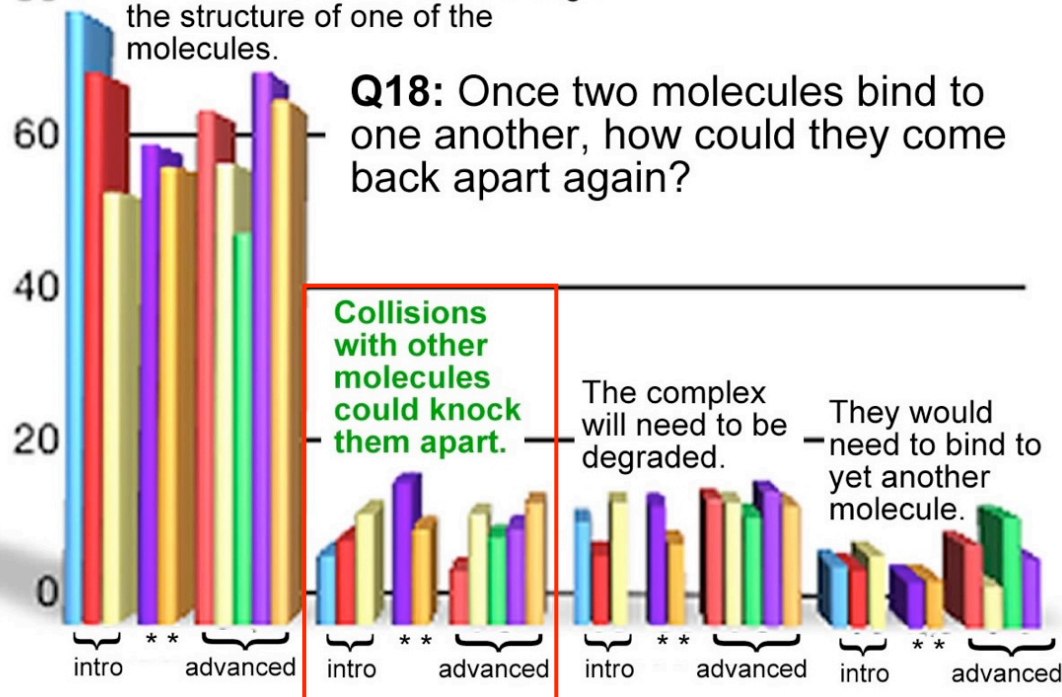
BCI Q25: Imagine an ADP molecule inside a bacterial cell. Which best describes how it would manage to "find" an ATP synthase so that it could become an ATP molecule?



Students prefer "active agents" over random movements

80 - A chemical reaction must change the structure of one of the molecules.

Q18: Once two molecules bind to one another, how could they come back apart again?



Garvin-Doxas & Klymkowsky. 2008. Understanding randomness and its impact on student learning: lessons learned from the Biology Concept Inventory (BCI). CBE Life Sci Educ 7: 227-233.

Klymkowsky, Furtak, Garvin-Doxas, Cooper & Gonzales, Understanding molecular creativity and evolutionary change: conceptual barriers, curricular omissions and didaskalogenic obstacles. in preparation.



Didaskalogenic (instructor/instruction-induced) confusion

Molecules interact through lock and key/
jig-saw puzzle (geometric) mechanism

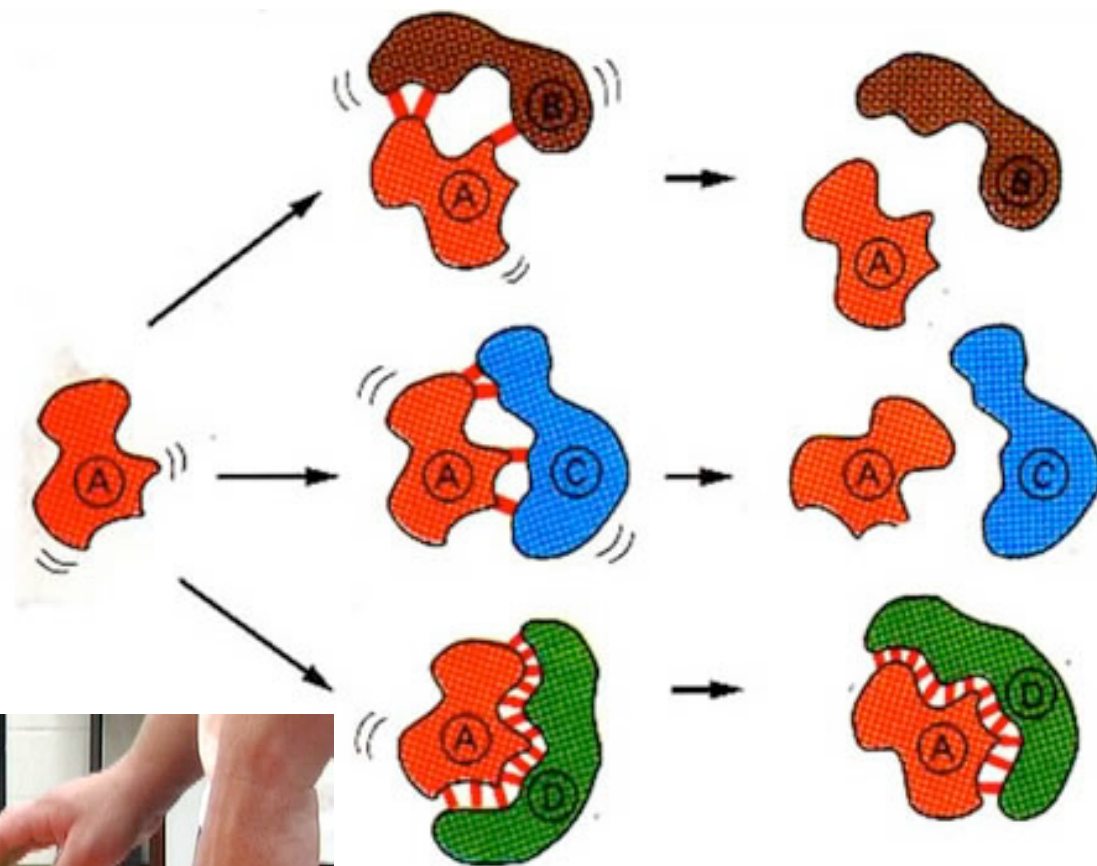
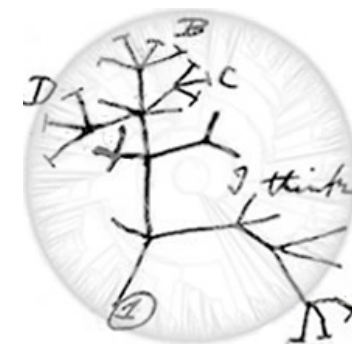
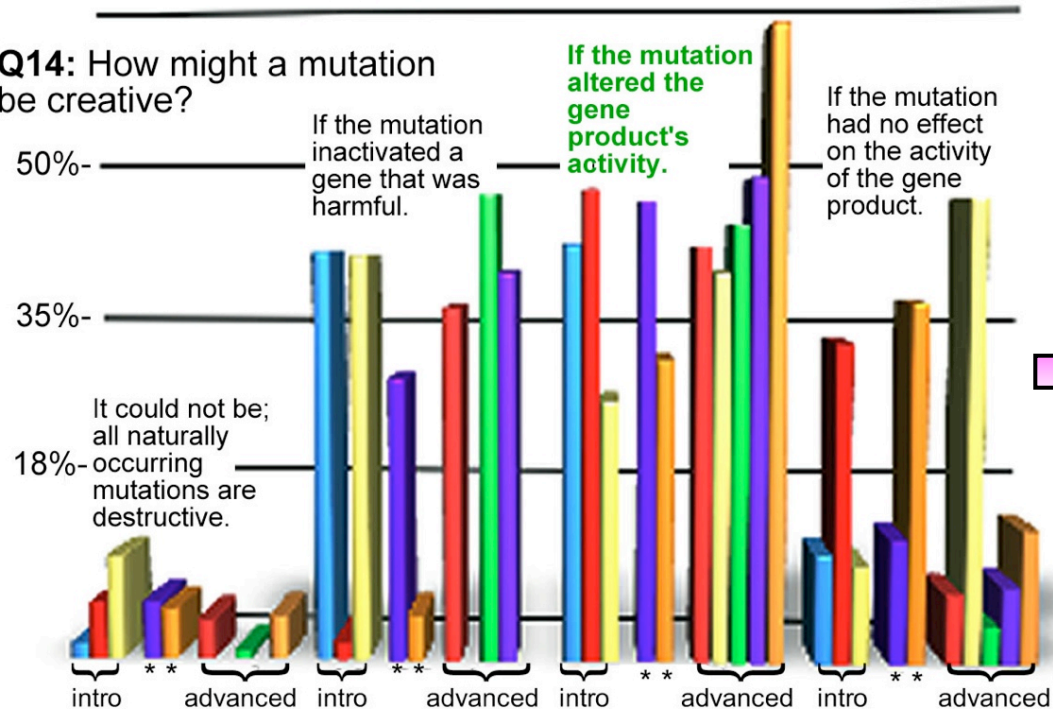


FIG. 1: Student working with FLOAM molecular models and their interactions during a "think-aloud interview"



Deeply held assumption, either immune to, or unaddressed by subsequent instruction

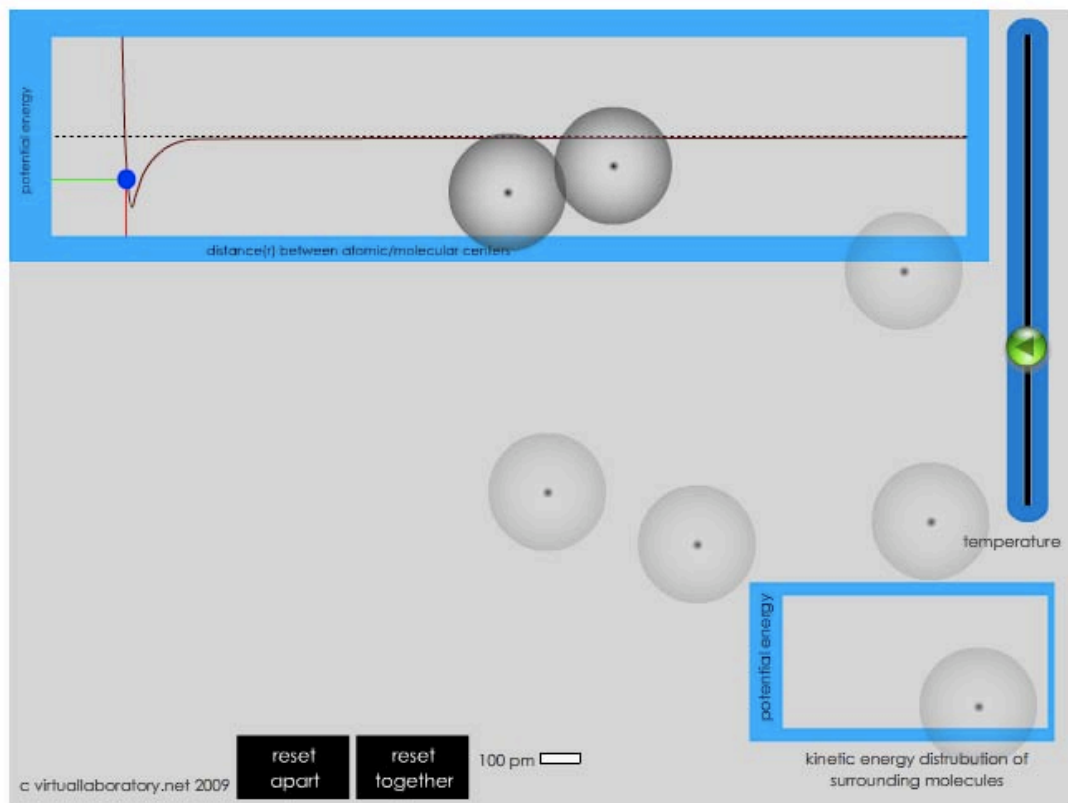
Q14: How might a mutation be creative?



Altered keys do not work.
Mutations are always destructive, never creative



Students need to be introduced to how molecules interact, how bonds form, and how systems behave.



What does the student see, what questions can they address, can they apply key concepts to new situations?



Students need test assumptions in new situations

potential energy

distance (r) between atomic/molecular centers

1) rotate and position

2) adjust temperature

3) analyze effects on interaction stability

temperature

kinetic energy distribution of surrounding molecules

reset apart

c virtuellaboratory.net

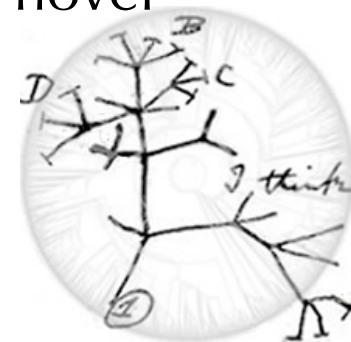
recognize:

1) the role of temperature as a determinant of stability

2) the reality of molecule promiscuity

3) the effects of mutations on interaction specificity, affinity, and activity

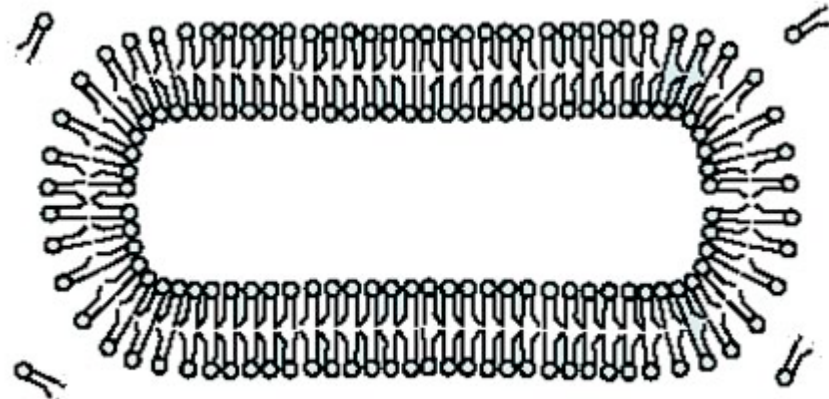
4) the need for gene duplication in the evolution of novel structures



Students need to think thermodynamically - systemically

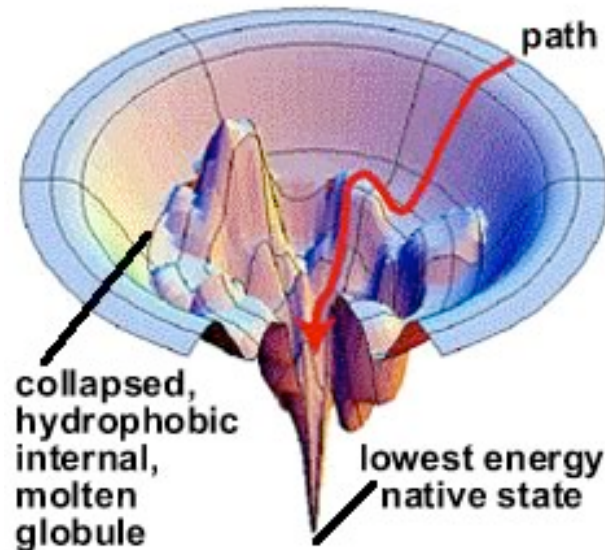
recognize that it is the differences between original and final state that matters (similar logic applies to energy release upon ATP hydrolysis)

increasing entropy drives the formation of ordered structures (origin of life)



entropy of lipid in water - entropy of bilayer in water

primary determinant of protein folding



These are concepts with a physical foundation: are they taught in the physics courses most biology, chemistry, and K12 science teachers take?

Has the content of first year physics been critically analyzed vis-a-vis its relevance to non-physicists? Is it pedagogically justified?

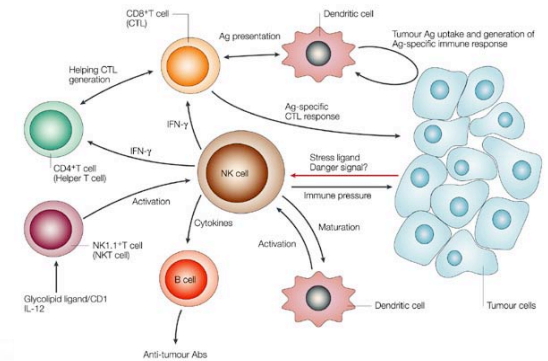


Life is about (evolving) networks

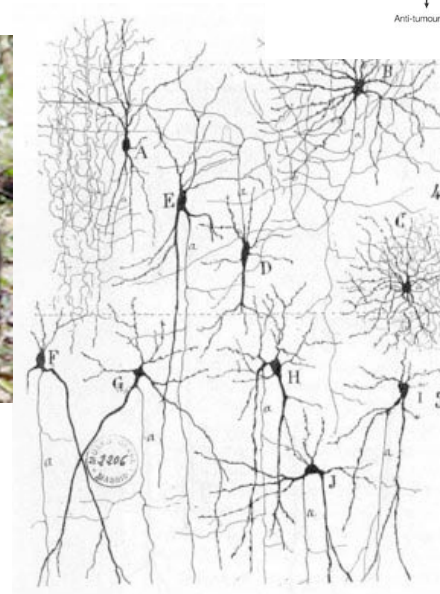
non-equilibrium, chemical reaction networks
gene regulatory networks

adaptive & homeostatic cellular networks
e.g. immune system / nervous system

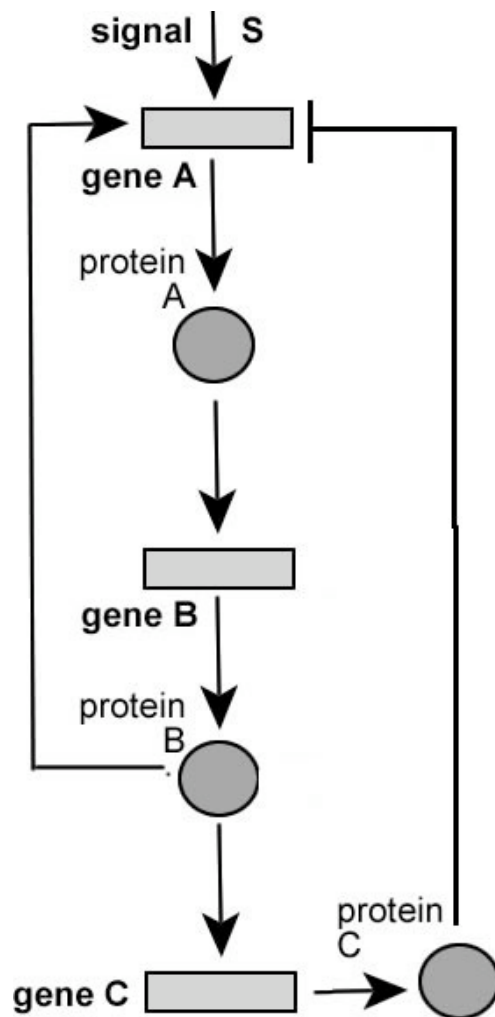
social networks
ecological networks



Nature Reviews | Cancer



physics / mathematics could help with ...

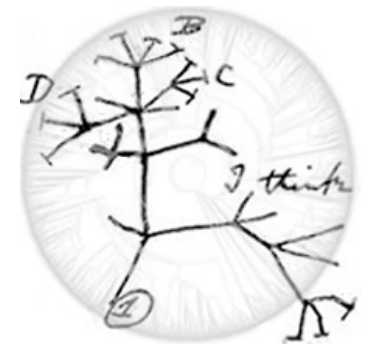


analytic analysis of network behaviors

particularly in the context of stochastic* processes (such as RNA/polypeptide half-lives under conditions of small number of molecules

Q: Are physics/math responsible for the mathphobic predilection of (most) biologists?

*student confusion as to whether this is property of individual molecules versus populations of molecules



Understanding biology (evolution) depends on a robust understanding of physical principles - no vital forces or designers needed.

Question is, should the required physical concepts be taught by physicists, or are you comfortable with biologists teaching them? The same applies you, chemists!



Acknowledgements:

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discipline based education research community @ UC Boulder

CU  Teach science/mathematic recruitment and certification program

NSF DUE: Building a basic biology concept inventory

NSF DUE: Chemistry and the logic of life

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